SIHS-N-189 A-21 B.Sc. III Semester (CBCS) Degree Examination CHEMISTRY

Paper : 3.1 (New)

Time: 3 Hours

Maximum Marks: 80

Instructions to Candidates:

Answer All the Three sections

SECTION - A

Answer All the following questions

 $(15 \times 1 = 15)$

- 1. What are Ideal solutions?
- 2. What is critical solution temperature?
- Define the term phase
- 4. What is ionic mobility?
- Define hydrolysis constant of a salt.
- 6. What are irreversible cells?
- Write nernst equation for Single electrode potential.
- Define standard electrode potential.
- 9. What are interhalogens?
- 10. What are amines?
- 11. Write the general formula of diazonium salt.
- 12. What is isoelectric point of amino acid?
- 13. What is Electrophoresis?
- 14. Write the ring structure of α -D-Glucose.
- 15. What is Mutarotation?

SECTION - B

11. Answer any Five of the following questions.

 $(5 \times 5 = 25)$

- 16. Explain phase diagram of two component system taking Lead- silver as an example.
- 17. Determine pH of a solution using hydrogen electrode.

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- 18. Derive Gibbs phase rule thermodynamically.
- 19. Give the preparation, properties and uses of interhalogen of the type AB,
- 20. How are aliphatic amines prepared from:
 - i) alkyl halides
 - ii) Hoffman's bromamide reaction?
- 21. What are proteins? Discuss the secondary structure of proteins.
- 22. Discuss the classification of carbohydrates with examples.

SECTION - C

111	4 -	ews=	Ony Five of the following and the	
111.			any Five of the following questions. (5×8:	
	23.	a)	Draw the vapour pressure-composition and boiling point-composition	
		L	curves.	(4)
		b)	State the following:	
			i) Raoult's Law	
	2.		ii) Lever rule	(4)
	24.	a)	Explain Moving boundary method for the experimental determination	
		L)	Transference number.	(4)
	25.	b)	Explain conductometric titration of a strong acid with a strong base.	(4)
	23.	a)	Explain the terms: https://www.karnatakastudy.com	
			i) EMF of a cell	
		L	ii) Electrochemical series	(4)
	26	p)	Write a note on Salt bridge	(4)
	26.	a)	Write a short note on Kohlrausch law of independent migration of ions	(4)
		ь)	Discuss the structure of diborane	(4)
	27.	a)	How do you distinguish primary, secondary and tertiary amines by ben sulphonyl chloride?	zene (4)
		b)	How will you convert diazonium into	(-)
			i) Phenol	
			ii) Benzene	(4)
	28.	a)	What are peptides? Give the synthesis of dipeptide.	(4)
		b)	Explain the following:	(4)
		-,	i) Strecker synthesis	
			ii) Gabriel phthalimide synthesis of amino acids	(4)
	29.	4)	Explain Ruff's degradation method for chain shortening of aldoses.	
,	27.	a)	- · · · · · · · · · · · · · · · · · · ·	(4)
		p)	Write the Haworth's structure for the following:	
		i) '''	Maltose	(4)
		ii)	Sucrose	(4)